CHEMISTRY MAJOR (B.S.)

https://ceps.unh.edu/chemistry/program/bs/chemistry-major

Description

The B.S. Chemistry degree is certified by the American Chemical Society and provides a deep, rigorous experience that prepares students for graduate work or a career in chemical industry and related fields. The curriculum offers thorough training in the major fields of chemistry, covering analytical, inorganic, organic, and physical chemistry, as well as biochemistry and chemical biology. Students gain laboratory experience in molecular synthesis and characterization, chemical biology, analytical and instrumental methods, physical chemical measurements and data analysis, and spectroscopy. At the same time, the program requires students to participate in scientific inquiry, via both advanced laboratory experiences and independent research.

Requirements

Degree Requirements

Minimum Credit Requirement: 128 credits

Minimum Residency Requirement: 32 credits must be taken at UNH

Minimum GPA: 2.0 required for conferral*

Core Curriculum Required: Discovery & Writing Program Requirements

Foreign Language Requirement: No

All Major, Option and Elective Requirements as indicated.

*Major GPA requirements as indicated.

Major Requirements

<table>
<thead>
<tr>
<th>Code</th>
<th>Title</th>
<th>Credits</th>
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<tbody>
<tr>
<td>CHEM 400</td>
<td>Freshman Seminar</td>
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<tr>
<td>CHEM 403</td>
<td>General Chemistry I</td>
<td>4</td>
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<td>CHEM 404</td>
<td>General Chemistry II</td>
<td>4</td>
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<tr>
<td>CHEM 517</td>
<td>Introduction to Chemical Measurement Science</td>
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<td>CHEM 518</td>
<td>Practical Chemical Measurement Techniques and Instrumentation</td>
<td>2</td>
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<tr>
<td>CHEM 547</td>
<td>Organic Chemistry I</td>
<td>5</td>
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<td>CHEM 549</td>
<td>Organic Chemistry Laboratory</td>
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<td>CHEM 548</td>
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<td>CHEM 550</td>
<td>Organic Chemistry Laboratory</td>
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<tr>
<td>CHEM 574</td>
<td>Chemistry Across the Periodic Table</td>
<td>6</td>
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<tr>
<td>CHEM 576</td>
<td>Experimental Inorganic Chemistry</td>
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<tr>
<td>BMCB 658</td>
<td>General Biochemistry 1; 3</td>
<td>3</td>
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<tr>
<td>or CHEM 740</td>
<td>Chemical Biology</td>
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<tr>
<td>CHEM 683</td>
<td>Physical Chemistry I</td>
<td>5</td>
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<tr>
<td>CHEM 684</td>
<td>Physical Chemistry II</td>
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<tr>
<td>CHEM 686</td>
<td>Physical Chemistry Laboratory</td>
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<tr>
<td>CHEM 755</td>
<td>Advanced Organic Chemistry</td>
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<tr>
<td>CHEM 762</td>
<td>Advanced Chemical Analysis Instrumentation</td>
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<td>CHEM 774</td>
<td>Inorganic Chemistry</td>
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<tr>
<td>CHEM 776</td>
<td>Physical Chemistry III</td>
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<tr>
<td>CHEM 777</td>
<td>Advanced Synthesis and Characterization</td>
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<tr>
<td>CHEM 798</td>
<td>Senior Seminar</td>
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<tr>
<td>CHEM 799</td>
<td>Senior Thesis 2</td>
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<tr>
<td>MATH 425</td>
<td>Calculus I</td>
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<tr>
<td>MATH 426</td>
<td>Calculus II</td>
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<tr>
<td>PHYS 407</td>
<td>General Physics I</td>
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BMCB 658 General Biochemistry satisfies the Discovery Biological Sciences requirement (for BS Chem majors only)

CHEM 799 Senior Thesis is a year-long experience of 4 credits per semester and satisfies the Discovery Capstone Experience requirement.

CHEM 740 Chemical Biology meets the ACS BIOCHEM requirement. Students can take CHEM 740 Chemical Biology. CHEM 740 Chemical Biology DOES NOT meet the Biological Sciences Discovery criteria. Students can meet this requirement with one of their discovery electives.

Degree Plan

This is the suggested degree plan for B.S. Chemistry majors. A student can alter this plan in consultation with an academic advisor.

Course | Title | Credits
--- | --- | ---
First Year | | |
Fall | | |
CHEM 400 | Freshman Seminar | 1
CHEM 403 | General Chemistry I | 4
MATH 425 | Calculus I | 4
PHYS 407 | General Physics I | 4
Discovery Course | 4

Credits | 17
Spring | | |
CHEM 404 | General Chemistry II | 4
MATH 426 | Calculus II | 4
ENGL 401 | First-Year Writing | 4
PHYS 408 | General Physics II | 4

Credits | 16
Second Year | | |
Fall | | |
CHEM 517 | Introduction to Chemical Measurement Science | 3
CHEM 518 | Practical Chemical Measurement Techniques and Instrumentation | 2
CHEM 547 | Organic Chemistry I | 3
CHEM 549 | Organic Chemistry Laboratory | 2
Discovery Course | 4
Discovery Course | 4

Credits | 18
Spring | | |
CHEM 548 | Organic Chemistry II | 3
CHEM 550 | Organic Chemistry Laboratory | 2
CHEM 574 | Chemistry Across the Periodic Table | 4
CHEM 576 | Experimental Inorganic Chemistry | 2
Discovery Course | 4

Credits | 15
Third Year | | |
Fall | | |
CHEM 683 | Physical Chemistry I | 3
CHEM 685  Physical Chemistry Laboratory  2
CHEM 774  Inorganic Chemistry  3
CHEM 755  Advanced Organic Chemistry  3
CHEM 777  Advanced Synthesis and Characterization  3
Discovery Course  4

Credits  18

Spring
CHEM 684  Physical Chemistry II  3
CHEM 686  Physical Chemistry Laboratory  2
CHEM 762  Advanced Chemical Analysis Instrumentation  3
CHEM 763  Advanced Chemical Instrumentation Laboratory  2
Elective Course  4
Discovery Course  4

Credits  18

Fourth Year

Fall
CHEM 776  Physical Chemistry III (Not offered every year. Can take CHEM 708 or CHEM 740 are Chem elective equivalent)  3
CHEM 799  Senior Thesis (first semester of a yearlong experience)  4
BMCB 658 or CHEM 740  General Biochemistry or Chemical Biology  3
Elective Course (1 course at 4 credits or 2 Electives at 4 credits each if not taking BMCB 658)  4

Credits  14

Spring
Elective Course  4
Elective Course  3
CHEM 798  Senior Seminar  1
CHEM 799  Senior Thesis  4

Credits  12

Total Credits  128

Student Learning Outcomes

• Reason with Chemistry’s anchoring concepts: that matter consists of atoms that have internal structures that dictate their chemical and physical behavior; that atoms interact via electrostatic forces to form chemical bonds that chemical compounds have geometric structures that influence their chemical and physical behaviors; that intermolecular forces—electrostatic forces between molecules—dictate the physical behavior of matter; that matter changes, forming products that have new chemical and physical properties that energy is the key currency of chemical reactions in molecular scale systems as well as macroscopic systems; that chemical changes have a time scale over which they occur; that all chemical changes are, in principle, reversible, and chemical processes often reach a state of dynamic equilibrium; that Chemistry is generally advanced via experimental observations; and that Chemistry constructs meaning interchangeably at the particulate and macroscopic levels.

• Use Chemistry’s cross-cutting concepts to interrogate and explain phenomena: chemical identity (how do we identify chemical substances?); structure-property relationships (how do we predict the properties of materials?); chemical causality (why do chemical processes occur?); chemical mechanism (how do chemical processes occur?); chemical control (how can we control chemical processes?); benefits-costs-risks (how do we evaluate the impacts of chemically transforming matter?).

• Demonstrate the following general scientific practices when displaying knowledge of chemical ideas and concepts: asking questions; developing and using models; constructing explanations; planning and carrying out investigations; engaging in argument from evidence; analyzing and interpreting data; using mathematics and computational thinking; obtaining, evaluating, and communicating information OR demonstrate the following Chemistry core practices when displaying knowledge of chemical ideas and concepts (a) analysis: development and application of strategies for detecting, identifying, separating, and quantifying chemical substances (b)synthesis: the design of new substances and synthetic routes (c)transformation: controlling chemical processes for non-synthetic purposes.